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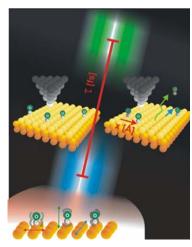
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tunneling microscopy (STM) is used to image the location and chemical identity of molecular species at a copper surface. Because STM has an								
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layers of the substrate, thereby creating temperature transients with hundreds of Kelvin amplitude but only picoseconds duration. We also use pores								
in adsorbate films to confine reactants.								
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Towards the Investigation of Molecular Dynamics at a Metal Surface Following Femtosecond Energetic Irradiation

Award No: FA9550-08-1-0401

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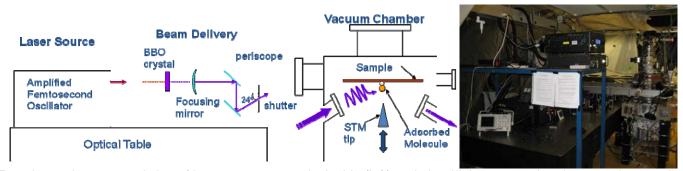
This project aimed at elucidating the dynamics of small molecules at metal surfaces at the atomic scale in a high-temperature environment. Scanning tunneling microscopy (STM) was used to image the location and chemical identity of molecular species at a copper surface. Because STM has an image acquisition time on the seconds to minutes scale, the sample generally needs to be cooled down to keep molecular adsorbates in place during imaging. A slight increase in temperature then allows the study of diffusion, but the adsorbates generally diffuse so fast that they cannot be studied anymore before other surface processes get into play. This leaves only molecular diffusion to be studied and it does not



Schematic representation of the anticipated experiments and temperature transient of substrate electrons, phonons and adsorbates following ultrafast substrate irradiation.

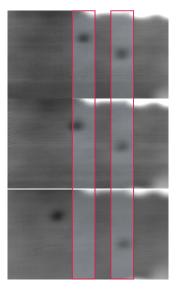
allow a realistic vista on what may occur at surfaces at elevated temperatures. To circumvent this obstacle, we use energetic femtosecond laser pulses to heat the top atomic layers of the substrate, thereby creating temperature transients with hundreds of Kelvin amplitude but only picoseconds duration [1]. Acquiring images of the same set of molecules before and after irradiation generates snapshots of surface dynamics only a few picoseconds in reaction time apart. Using corrals created from anthraquinone, we locally confine adsorbates

To this end a system combining a powerful femtosecond laser system and a low temperature scanning tunneling microscope has been assembled.



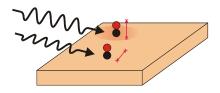
Experimental setup consisting of laser system on optical table (left) and ultra-high vacuum chamber contain scanning tunneling microscope (right)

Our measurements addressed the diffusion of carbon monoxide molecules on the Cu(111) surface. Working at 20K, i.e. well below the native diffusion threshold of 35K, we observe ample diffusion of adsorbed species following irradiation with 3.4mJ/cm² at 400nm and 100fs. To correlate the measurements with the energy redistribution in the surface, we currently acquire a dataset, in which we compare the threshold to obtain molecular motion between s and p polarization of the incident irradiation. Interaction of the metal surface with irradiation of different polarization planes is fundamentally different: while the field of an electromagnetic wave oscillating plane parallel to the surface can be compensated by rearrangement of the electrons in the surface plane leading to reflection, the same is not possible for a wave polarized



Diffusion of CO molecules on Cu(111) following irradiation with 100fs, 3.4 mJ/cm², 400nm pulses

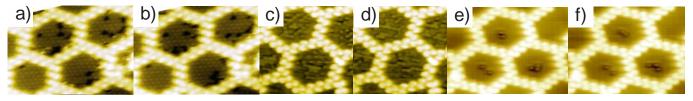
perpendicular to the surface. Rather, such a wave is aligned with the molecular axis and can, potentially, couple more directly and athermally to the molecule.



Our measurements show that the amount of molecular excitation is not increased for p polarization, thus affirming that the molecular motion we induce in our experiments is indeed a thermal effect and validating that the methods is

sutiable to provide us a vista on the high-themperature molecular dynamics at the focus of this proposal.

To facilitate tracking of the molecules addressed, we developed a technique that involves molecular corrals formed by anthraquinone adsorbates on Cu(111)[2]. The pores of this network are capable of confining CO molecules within a very small area, thus preventing that they run across extended terraces. This allows following tens of events for one molecule[3].



Different number (a,b: ~60; c,d: ~20; e,f: 2-3) of CO molecules confined in anthraquinone networks and their diffusion.

References:

- 1. Bartels, L., et al., *Real-space observation of molecular motion induced by femtosecond laser pulses*. Science, 2004. **305**(5684): p. 648-651.
- 2. Pawin, G., et al., *A homomolecular porous network at a Cu(111) surface*. Science, 2006. **313**(5789): p. 961-962.
- 3. Cheng, Z., et al., *The Power of Confinement: Adsorbate Dynamics on Nanometer-Scale Exposed Facets.* NANO LETTERS, 2010. **10**: p. 1022018